Appendix C Example Science Items

Example Science Items Grade 4



The figure **above** shows a box that contains a material that could be a solid, a liquid or a gas. The material is then put into a box four times as large.

Look at the figures **below**. They show how the different types of material will look when put into the larger box.

A. Identify which figure shows a solid, which shows a liquid and which shows a gas. (Write the word *solid*, *liquid* or *gas* on the line next to each figure below. Use each word only once.)

B. Explain your answers.

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Liquid is runny so it finds the lowest place in the box.

945

A solid can be everywhere in the box but it stays the same shape.

A gas takes up all the room it wants.







Name of Organism	What the Organism Needs for Food	
Seaweed	Sunlight to make its own food	
Limpet	Seaweed	
Crab	Limpets	
Octopus	Limpets, crabs, and fish	
Seagull	Crabs and fish	
Seal	Crabs, octopus and fish	

Questions for Ocean Food Chain begin on the next page.

The diagram below shows part of a food chain. The arrows go from one organism to another organism that eats it. In this food chain, the limpets eat seaweed.



- A. Complete the food chain **above** by writing the names of two other organisms from the table in the blank spaces. Use the information in the table about what each organism needs for food. (*There is more than one correct food chain. You need to show just one.*)
- B. One year a disease causes many limpets to die. What would happen to the **seaweed** in your food chain when the limpets die?

The amount of seaweed will grow because there is not lots of limpets to eat it.

C. Choose another organism in your food chain (not seaweed or limpet). Name of organism: **Crab**

What would happen to this organism when the limpets die?

Some crabs will die because there isnt a lot of limpets to eat.

D. What would happen to the other organisms in your food chain if the seaweed does not grow well?

The limpets will starve, so some crabs will die, so some octopuses will die. If one tiny thing happens to an animal or plant, it can affect the whole food chain.

End of Ocean Food Chain questions.

Example Science Items Grade 8





The picture shows how a student set up some apparatus in a laboratory for an investigation. The inverted test tube was completely filled with water at the beginning of the investigation as shown in Figure 1. After several hours, the level of water in the test tube had gone down as shown in Figure 2.



What is contained in the top part of the test tube labeled X in Figure 2?

(Check one box.)

3

air air

carbon dioxide

vacuum

Explain your answer.

During photosynthesis, plants produce oxygen and glucose.

Appendix C: Example Science Items – Grade 8



Three identical candles are placed in the three jars shown above and lit at the same time. Jars Y and Z are then sealed with lids, and Jar X is left open.

Which candle flame will go out first (X, Y, or Z)?

Explain your answer.

Z because fire needs oxygen to stay lit. With the lid being sealed no oxygen can get in. There is a little bit of air in these for it to stay lit. Since Z is smaller than Y, Z would go out first.





Metal Crown Instructions: Questions 7, 8, 9, 10 are about Metal Crown. To answer these questions you may refer to any information

shown on the pages in the Metal Crown section.

A king gave a jeweler a block of pure metal. He asked the jeweler to make him a crown out of the metal.





After the jeweler delivered the crown, the king observed it carefully. He thought that the jeweler might have substituted another pure metal or a mixture of metals to make the crown. He weighed the crown, and it had the same mass as the original block, 2400 grams. Still not satisfied, the king asked some scientists to help him find out what the crown was made of.

Questions for Metal Crown begin on the next page.



The scientists decided to compare the densities of the crown and a block of metal just like the originalblock. The density of a substance is the mass of a sample of the substance divided by its volume (density = mass/volume).

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The scientists found the volume of the block and computed its density based on its known mass (2400g). The diagram below shows the dimensions of the block of metal that the scientists measured.

(not to scale) 5 cm	5
What is the density of the block of metal?	
Answer: 19.2 g/cm ³	
This tenner extron	

Questions for Metal Crown continue.

The scientists then needed to find the volume of the crown in order to determine its density. The following equipment and materials were available for them to use.



Describe a procedure that the scientists could use to find the volume of the crown using some or all of the equipment and materials shown above. You may use diagrams to help explain your procedure.

fill the beaker with enough water to cover the crown. Add the crow and mark the side of the beaker where the water level is . Then take the crown out. Use the graduated cylinder to add little bits of water Until the level comes back up to the mark. That is the Volcime of the Crown.

Questions for Metal Crown continue.

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The scientists measured the volume of the crown five times. They computed the density for each volume measurement. Their results are shown in the table below.

	Trial	Volume of Crown (cm ³)	Density of Crown (g/cm ³)
	1	202	11.88
	2	200	12.00
	3	201	11.94
	4	198	12.12
X	5	199	12.06
<u> </u>			

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A. Why did the scientists measure the volume five times?

Because there is experimental error. 50, me asuring it 5 times you can calculate the overage to know how much error there i

B. The scientists reported to the king that the density of the crown was 12.0 g/cm³. Show how the scientists used their results to obtain this value for the density.

They added together all of the densities and then divided by 5 to get the average.

Questions for Metal Crown continue.

The table below lists the density for different metals.

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Metal	Density (g/cm³)
Platinum	21.4
Gold	19.3
Silver	10.5
Copper	8.9
Zinc	7.1
Aluminum	2.7

10

A. Look at the density you computed for the block of metal. What was the block of metal most likely made of?

closest

Answer: 601d

Explain your answer.

It had the

B. The density of the crown was found to be 12.0 g/cm³. What would you report to the king about what metal or mixture of metals the jeweler used to make the crown?

The jeweler used some silver as well as gold.

densi

